

Manipulating the Amyloid- β Aggregation Pathway with Chemical Chaperones*

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Amyloid- β ($A\beta$) assembly into fibrillar structures is a defining characteristic of Alzheimer's disease that is initiated by a conformational transition from random coil to β -sheet and a nucleation-dependent aggregation process. We have investigated the role of organic osmolytes as chemical chaperones in the amyloid pathway using glycerol to mimic the effects of naturally occurring molecules. Osmolytes such as the naturally occurring trimethylamine *N*-oxide and glycerol correct folding defects by preferentially hydrating partially denatured proteins and entropically stabilize native conformations and polymeric states. Trimethylamine *N*-oxide and glycerol were found to rapidly accelerate the $A\beta$ random coil-to- β -sheet conformational change necessary for fiber formation. This was accompanied by an immediate conversion of amorphous unstructured aggregates into uniform globular and possibly nucleating structures. Osmolyte-facilitated changes in $A\beta$ hydration also affected the final stages of amyloid formation and mediated transition from the protofibrils to mature fibers that are observed *in vivo*. These findings suggest that hydration forces can be used to control fibril assembly and may have implications for the accumulation of $A\beta$ within intracellular compartments such as the endoplasmic reticulum and *in vitro* modeling of the amyloid pathway.

Amyloid plaques are a central feature of Alzheimer's disease pathology and are considered to be a major factor in neuronal cell loss (1, 2). Ultrastructurally, plaques are fibrous masses composed primarily of the 40–42-residue amyloid- β ($A\beta$)¹ peptide. $A\beta$ is derived by endoproteolysis of the integral membrane amyloid precursor protein that results in secretion of the peptide by normal cellular pathways as well as intracellular accumulation. Under pathogenic conditions, $A\beta$ self-associates into

a well defined supramolecular fibril with high β -sheet content (3). $A\beta$ polymerization is considered to be a two-stage process initiated by the association of individual $A\beta$ monomers into small nucleating "seeds" that is accompanied by a transition from a predominately random coil to an amyloidogenic β -sheet conformation (4, 5). Subsequent to nucleation, the $A\beta$ seeds assemble in a chain-like manner to yield an intermediate protofibrillar structure (6–8), which may represent a common element of all amyloid fibrils (9, 10). Protofibril conversion into the ramified fibrils observed *in vivo* can be affected by factors such as the amyloid-binding apoE, which is an Alzheimer's disease risk factor (11); the relative quantity of the more amyloidogenic $A\beta$ species (4); and other chaperone elements that may control $A\beta$ self-association.

The observation that $A\beta$ is generated within intracellular compartments, including the endoplasmic reticulum (12–15), which is the quality control site for protein folding, has prompted us to investigate the role of chemical chaperones in the $A\beta$ folding pathway. Under stress conditions or exposure to denaturants, heat shock proteins and chemical chaperones assist in stabilizing correctly folded proteins. Through a process known as osmotic remediation (16), chemical chaperones or organic osmolytes, including carbohydrates, free amino acids, or methylamines (*e.g.* trimethylamine *N*-oxide (TMAO)), effectively control protein folding through a preferential hydration of exposed polypeptide backbone and side chains of partially unfolded structures (17). Chemical agents such as glycerol and polyethylene glycol mimic these hydration effects, which creates a thermodynamically unstable state due to the unfavorable entropic changes associated with the increased ordering of bound water molecules (18). This can be rectified by folding of the protein into its native conformation, which sequesters the exposed groups and excludes the osmolytes from the protein domain. As a result, the free energy of the native conformation is substantially lower than the unfolded state, which is demonstrated by the ability of glycerol to correct misfolded proteins within the endoplasmic reticulum (19). Hydration effects are equally important in protein polymerization where osmolytes are excluded through increased protein-protein interactions and can, for example, enhance the assembly and stability of microtubules (20). The preferential assembly of protein polymer subunits such as $A\beta$ and tubulin is a product of their unique structure, which results in the fibrous aggregate being the lowest energy conformer. Our findings indicate that similar forces contribute to the initiation of the $A\beta$ random coil-to- β -sheet transition and stabilization of the resulting aggregates. These observations suggest that chemical chaperones may be useful in modeling amyloid plaque formation and may have some bearing on the cellular events involved in fibril formation.

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¹ The abbreviations used are: $A\beta$, amyloid- β ; TMAO, trimethylamine *N*-oxide; NBD, *N'*-(7-nitrobenz-2-oxa-1,3-diazol-4-yl)ethylenediamine; TMAFM, tapping mode atomic force microscopy; AFM, atomic force microscopy.

EXPERIMENTAL PROCEDURES

Peptide Synthesis and Labeling—Amyloid- β peptide residues 1–40 (A β 40) was synthesized and purified as described previously (21). Peptides were examined immediately following dissolution in aqueous buffer and following prolonged incubation at high concentration to promote small protofibrillar β -sheet aggregates. A β 40 was labeled for fluorescence studies using *N,N'*-dimethyl-*N*-(iodoacetyl)-*N'*-(7-nitrobenz-2-oxa-1,3-diazol-4-yl)ethylenediamine (Molecular Probes, Inc.) as described (22). *N'*-(7-Nitrobenz-2-oxa-1,3-diazol-4-yl)ethylenediamine (NBD)-labeled peptide was purified by gel filtration chromatography and disaggregated in 10% hexafluoro-2-propanol, and stock solutions were stored at -20°C .

Secondary Structure Analysis—The effects of TMAO (Sigma) and glycerol (BDH) on A β conformation were determined by CD. Peptides were dissolved to a final concentration of $50\ \mu\text{M}$ in distilled water or $10\ \text{mM}$ phosphate buffer (pH 7). Peptide solutions were combined with the TMAO at concentrations ranging from 50 to $150\ \mu\text{M}$ and with glycerol ranging from 1.2 to $6\ \text{M}$ (10–50% by volume). Spectra were collected following a 10-min equilibration period and after 48–72 h of incubation at room temperature. Spectra were acquired on a Jasco Model J-715 spectropolarimeter in a 0.1-cm path length cell over a wavelength range of 190 – $250\ \text{nm}$ with a 1.0-nm bandwidth, 0.1-nm resolution, 1-s response time, and $20\ \text{nm}/\text{min}$ scan rate. All spectra were corrected by subtraction of any contributions from buffer, glycerol, TMAO or polyethylene glycol.

Electron and Atomic Force Microscopy—A β aggregates were examined by phosphotungstic acid negative staining and platinum/carbon shadowing techniques as described previously (17). For shadowing studies, the samples were atomized onto freshly cleaved mica, immediately plunged into liquid nitrogen, and lyophilized to eliminate drying artifacts that could be caused by changes in peptide concentration. Dried preparations were platinum-coated in an Edwards E12E4 coater and viewed on a Hitachi H-7000 electron microscope operated with an accelerating voltage of $75\ \text{kV}$.

For tapping mode atomic force microscopy (TMAFM) studies, the peptides were dissolved in $25\ \text{mM}$ phosphate buffer (pH 7) and then adjusted to the desired TMAO or glycerol content to a final peptide concentration of $2.5\ \mu\text{M}$. Approximately $10\ \mu\text{l}$ of the peptide solution was transferred to a piece of freshly cleaved mica glued to a steel AFM sample mount. The sample was then immediately sealed in the TMAFM liquid cell, and the cell was filled with buffer solution. TMAFM imaging was conducted at room temperature using a combination contact/tapping mode liquid cell fitted to a Digital Instruments Nanoscope IIIA MultiMode scanning probe microscope. All images were acquired using $120\text{-}\mu\text{m}$ oxide-sharpened silicon nitride V-shaped cantilevers (type DNP-S, Digital Instruments Inc., Santa Barbara, CA) at a scan rate of $\sim 2\ \text{Hz}$ and a sampling rate of 256 or 512 points/scan line. Prior to use, the AFM tips were exposed to UV irradiation to remove adventitious organic contaminants from the tip surface. While *a priori* determination of the appropriate drive frequency is difficult owing to viscous coupling between the cantilever and the fluid medium, which gives rise to multiple broad resonance peaks, optimal imaging was achieved at a cantilever drive frequency of $\sim 8.9\ \text{kHz}$.

Solubility Measurements—The aggregation state of A β under the various solvent conditions was examined by high speed centrifugation and assay of the soluble material using the fluorescently labeled peptide as indicator. Solutions containing $0.1\ \mu\text{M}$ NBD-labeled A β 40 and $10\ \mu\text{M}$ unlabeled peptide were combined in $25\ \text{mM}$ phosphate buffer (pH 7) containing from 0 to $6.0\ \text{M}$ glycerol. NBD-labeled A β 40 fluorescence spectra were acquired; the samples were centrifuged in a Beckman Airfuge at maximum velocity ($135,000 \times g$) for $30\ \text{min}$; and the fluorescence spectra of the supernatants were measured. Spectra of control samples containing only A β 40 were collected and subtracted from the NBD-labeled A β 40 fluorescence to correct for the effects of light scattering. Steady-state fluorescence was measured at room temperature using a Photon Technology International QM-1 fluorescence spectrophotometer equipped with excitation intensity correction. Emission spectra from 500 to $600\ \text{nm}$ were collected ($\lambda_{\text{ex}} = 478\ \text{nm}$, $0.1\ \text{s}/\text{nm}$, 8-nm band pass for excitation and emission) using a $0.2 \times 1\text{-cm}$ path length, 0.5-ml cuvette. To determine the effects of glycerol on the fluorescent probe, unconjugated NBD at $0.1\ \mu\text{M}$ was measured under comparable conditions.

RESULTS AND DISCUSSION

Circular dichroism was used to evaluate the effects of the naturally occurring osmolyte TMAO and glycerol on A β conformation. When dissolved in aqueous buffers, A β 40 initially ex-

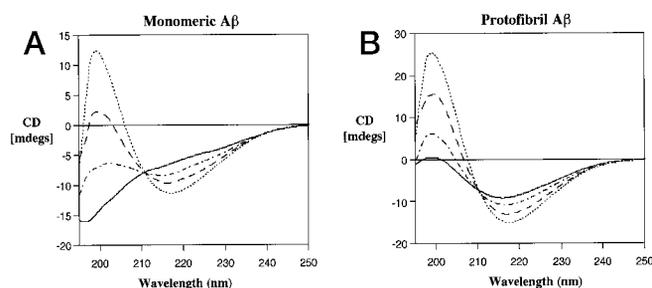


FIG. 1. A, CD of A β 40 demonstrating the immediate conversion from random coil to a β -sheet conformation with increasing glycerol concentrations; B, conformational transitions of an incubated or “aged” A β 40 comparable to the protofibril state indicating a similar effect of increasing the β -sheet content induced by glycerol-mediated solvation effects. Spectra are of the A β peptide in buffer (—) and in the presence of $1.2\ \text{M}$ (---), $3.0\ \text{M}$ (-·-), or $6.0\ \text{M}$ (···) glycerol. *mdeg*s, millidegrees.

hibits a random coil conformation indicative of an unordered structure (Fig. 1A). Consistent with previous reports, the conformational changes of A β from random coil to β -sheet (21) and subsequent fibril formation (23) can take from hours to days depending on the particular peptide batch and incubation conditions. In contrast, adjusting the solution to $1.2\ \text{M}$ glycerol (10%, v/v) resulted in an immediate folding of the peptide into the amyloid-associated β -conformer. Increasing the glycerol level to 3 or $6\ \text{M}$ (25–50%, v/v) produced a linear increase in the β -sheet content as measured by the minima intensity at $218\ \text{nm}$ (Fig. 1A). These effects were not due to increased peptide concentration caused by immiscibility in glycerol since CD studies conducted in which A β concentrations were doubled (from 50 to $100\ \mu\text{M}$) to replicate the 50% glycerol conditions revealed no change in the random coil structure of A β 40 (data not shown). A similar change was observed for A β 40 that had been preincubated to form β -sheet aggregates, which represent the early stages of fibril formation. In this case, elevating glycerol concentrations proportionally increased the preexisting β -sheet content, but to a slightly lesser quantitative degree as compared with the random coil-to- β -sheet conformational change (Fig. 1B). Identical results were obtained under all conditions in the presence of a low molecular mass ($400\ \text{Da}$) polyethylene glycol at varying concentrations (data not shown). These results indicate that changes in protein hydration by chemical chaperones rapidly accelerate the conformational transition required for amyloid formation.

Glycerol is a nonphysiological model of osmolyte activity, and we therefore investigated the effect of TMAO, which is found *in vivo* and acts to maintain correctly folded proteins in several species (17). TMAO produced a similar random coil-to- β -sheet transition as that seen with glycerol, but at significantly lower concentrations of 50 – $150\ \mu\text{M}$ (corresponding to molar ratios of 1:1 and 1:3 peptide/TMAO) (Fig. 2A). Similar to glycerol, TMAO increased the quantity of the β -conformation with the preincubated peptide, which initially displayed a folded and aggregated structure (data not shown). The increases in β -sheet conformation by both TMAO and glycerol were found to be roughly linear as determined by the absorption at $218\ \text{nm}$ (Fig. 2B). However, at the higher concentrations of TMAO, the proportion of the β -conformation appeared to be approaching a plateau. These findings suggest that TMAO is a more active compound in controlling the folding and aggregation state of A β , which may reflect a more potent effect of this osmolyte in an *in vivo* setting.

The morphological changes associated with the β -sheet conformation were assessed by TMAFM performed directly in aqueous buffer. TMAFM of A β in the absence of the chemical chaperones revealed irregular aggregate rafts with an approx-

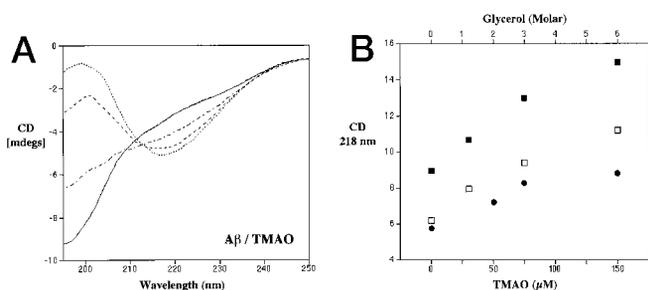


FIG. 2. *A*, CD of A β 40 demonstrating the immediate conversion from random coil to a β -sheet conformation with increasing concentrations of TMAO. Control A β in buffer only (—) and A β /TMAO at molar ratios of 1:1 (---), 1:1.5 (---), and 1:3 (···) are indicated. *B*, proportional increases in the β -sheet conformation as measured by the absorption at 218 nm in the presence of TMAO (●) and glycerol (□) with A β initially in the random conformation and following preincubation prior to the addition of glycerol (■). *mdeg*s, millidegrees.

imate thickness of 10 nm (Fig. 3A). The presence of aggregates was unexpected since it is generally assumed that the random coil conformation corresponds to a fully solvated and soluble A β monomer. Since the samples were not pretreated to remove small peptide aggregates (*e.g.* submicron filtration), these minor components could be present and serve as nuclei for non-specific precipitation. Alternatively, the amorphous deposits may be due to adsorption of A β to the mica substrate used for TMAFM. The addition of either TMAO or glycerol at concentrations that produced the random coil-to- β -sheet transition, as confirmed by CD, resulted in a complete conversion of the amorphous A β aggregates to a mixture of protofibrils and small ellipsoidal particles (Fig. 3, *B* and *C*). The protofibrils varied in length and displayed a globular axial periodicity of ~ 100 Å that was comparable to their average diameter. These were morphologically similar to previously reported protofibrils (7, 8). However, judging from wider field scans, the ellipsoidal aggregates predominated with dimensions of $50 \times 60 \times 15$ Å (Fig. 3C, *arrow*). We noted that these dimensions are slightly overestimated due to convolution of the AFM tip shape with the aggregate shape.

Given an experimentally determined volume of $\sim 45,500$ Å³, these ellipsoid particles represent A β tetramers or pentamers based on a calculated volume of $\sim 10,000$ Å³ for an A β (residues 1–42) molecule folded into a two-stranded β -sheet. These may represent an aggregate of the A β dimer that has been shown to be stable under similar aqueous conditions (24). Comparable structures have been observed by AFM (25, 26) and have been termed A β -derived diffusible ligands, which may represent the most neurotoxic species (27). The position of these small aggregates on the amyloid pathway is presently unclear. A straightforward explanation would be that there is a linear relationship where these are the progenitors of the protofibrils that are generated by the direct polymerization of the ellipsoid aggregates. Alternatively, they may represent a side reaction with A β monomers shuttling between these and the protofibrils.

Assembly of the nucleating aggregates to form protofibrils is considered to be the slow kinetic phase of the amyloid pathway (28). This is followed by the thermodynamic phase with the transition to compacted amyloid fibrils, a process that may also be affected by osmolyte-induced hydration. Previous AFM and negative stain electron microscopy studies have defined protofibrils as truncated and highly flexible fibrillar structures that are the precursors to plaque-related fibrils (6, 7, 8). Following preincubation of a concentrated A β 40 solution, we have observed similar aggregates of similar morphology using platinum/carbon shadowing techniques (Fig. 4A). Such protofibrils ranged in length from 1 to 100 nm and were poorly contrasted, suggesting that they have a low profile. The addition of glycerol

resulted in a rapid conversion to straight compacted fibrils that extended over several hundred nanometers (Fig. 4B) and occasionally displayed helical twisting (*arrowheads*). Exposure to chemical chaperones such as glycerol also significantly improved the image contrast of the fibrils, presumably as a result of their compaction into more defined tubular assemblies. Although some curved protofibril-like structures remained following glycerol treatment, there was a complete conversion to the longer fibrils following several hours of incubation (data not shown). Elongation occurred at a significantly greater pace as compared with the several days required to convert A β protofibrils in the absence of the glycerol chaperone.

To obtain a quantitative measure of aggregation under our experimental conditions, centrifugation was employed using A β labeled with the fluorescent probe NBD. Our previous fluorescence resonance energy transfer studies have indicated that the low concentrations of the fluorophore-labeled A β do not interfere with the kinetics of fibril formation or the morphology of the resulting aggregates (22). In the current study, similar peptide solutions containing excess unlabeled (10 μ M) and NBD-labeled (0.1 μ M) A β were used. Prior to centrifugation, the fluorescence intensity increased linearly with increasing concentrations of glycerol, with the fluorescence at 6 M glycerol being approximately twice that of the buffer-only sample (Fig. 5). This increase was due to glycerol-induced changes in the quantum yield of NBD as shown by the response of the unconjugated label under these conditions. Following centrifugation, a 97% decrease in fluorescence was observed, indicating that the peptides are almost completely aggregated. This was the case for the glycerol-containing samples as well as the initial aqueous solution, which is consistent with the amorphous aggregates observed by AFM. This is likely due to the fact that the samples were not pretreated, for example, by filtration to remove nonspecific aggregates, which would have allowed examination of the more direct conversion of soluble monomer to A β aggregate. However, the results presented here with untreated samples suggest that the conversion to the β -sheet particles, which are induced by solvation changes, may involve a shuttling of monomers from the unordered structure to the nucleating aggregates. An analogous situation may occur with the diffuse-to-senile plaque A β interconversion that is seen *in vivo*, and the use of chemical chaperones may be useful in modeling this aspect of the amyloid pathway.

Cumulatively, these results demonstrate that changes in the solvation state of the A β peptide affect several aspects of the amyloidogenic pathway. In general terms, amyloid fibrils are initiated by the destabilization of a normal cellular protein that leads to a partially unfolded intermediate. This can be accelerated by point mutations, as has been shown for lysozyme (29) and transthyretin (30), which are deposited as systemic plaques. If uncorrected, the unfolded intermediate will ultimately assemble into the β -sheet aggregates that initiate fibril formation. It has been proposed that chemical chaperones may, in part, regulate amyloid formation by stabilizing the lower energy native conformer to reduce the levels of unfolded proteins that are required for the amyloidogenic pathway. This is consistent with the observation that naturally occurring organic osmolytes inhibit the conversion of the cellular prion protein (PrP^C) to the protease-resistant and amyloid-forming PrP^{Sc} associated with transmissible encephalopathies (31, 32). Similar stabilization by glycerol of misfolded mutant transmembrane proteins within the endoplasmic reticulum has been demonstrated for the chloride transporter associated with cystic fibrosis (33, 34) and aquaporins related to nephrogenic diabetes (35).

Although stabilizing native conformations of larger proteins

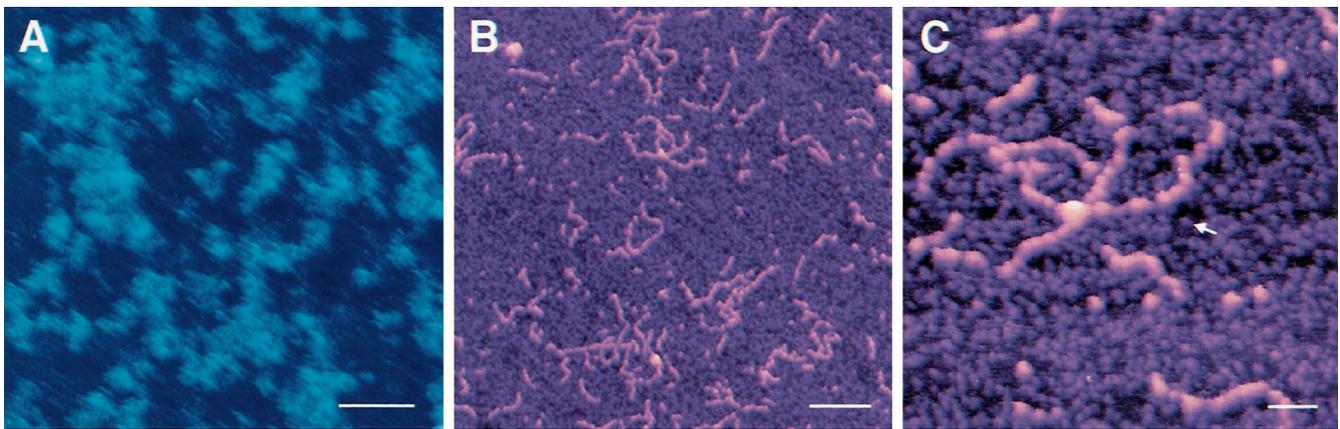


FIG. 3. **Morphological changes in peptide aggregates induced by chemical chaperones.** A, atomic force microscopy of A β 40 (2.5 μ M) in phosphate buffer demonstrating the amorphous aggregates. B, a wide field scan of an identical A β 40 peptide in 6.0 M glycerol revealing the formation of protofibrils and smaller nonfibrillar aggregates. Identical structures were observed in the presence of 150 μ M TMAO. C, enlarged view of the protofibrils indicating the globular morphology and the discrete ellipsoidal aggregates (arrow) representing the tetrameric/pentameric aggregate. Scale bars are 100 nm in A, 2500 nm in B, and 500 Å in C.

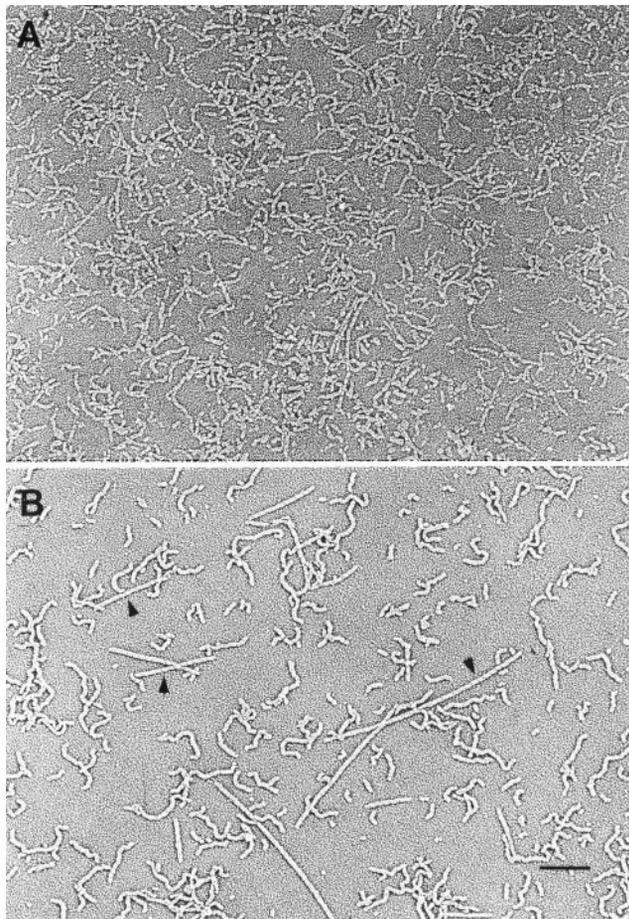


FIG. 4. A, rotary platinum/carbon shadowing electron microscopy of A β 40 demonstrating the presence of protofibrils in a preincubated or aged preparation; B, transition to amyloid fibrils in 6.0 M glycerol showing the formation of elongated and rigid fibers that occasionally exhibited a helical twisting (arrowheads). Scale bar is 1000 Å.

can affect amyloid formation, it is likely that A β does not have a conventional conformation and may therefore exist in an unfolded and possibly amorphous state. Therefore, if unchecked by cellular control elements, the peptide will gradually deposit as diffuse and eventually senile plaques. Our results indicate that solvation changes induced by chemical chaperones such as the naturally occurring TMAO and the *in vitro* model provided by glycerol rapidly accelerate the major steps in

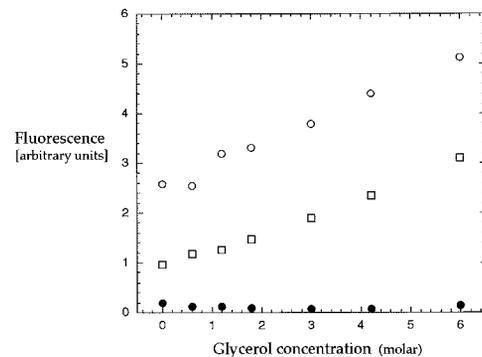


FIG. 5. **Insolubility of the unstructured and fibrillar aggregates monitored by a fluorescent labeled A β peptide tracer.** Peptides incubated in aqueous phosphate buffer or with increasing concentrations of glycerol were subjected to high speed centrifugation. Precipitation of peptide was observed in all cases as shown by the loss of fluorescence observed in the initial solutions (○) as compared with the supernatant fluorescence following centrifugation (●). The increase in fluorescence at higher glycerol concentrations is due to changes in the probe as shown by similar changes in un conjugated label (□).

the amyloidogenic pathway. These include both the early nucleation and conformational events as well as the protofibril-to-fiber conversion. This may be unique to proteins that normally exist in polymeric forms as shown by chemical chaperone-mediated assembly of tubulin into microtubules and their subsequent resistance to urea denaturation (20). The significance of our findings is that they reveal additional contributors to A β fibril formation and therefore provide an additional tool to manipulate this pathway. Such information could potentially be used to develop or accelerate cellular models of A β aggregation and the assessment of agents that modulate fibril formation.

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